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(54) PESTICIDAL PYRIMIDINE COMPOUNDS

PESTIZIDE PYRIMIDIN VERBINDUNGEN

COMPOSES PESTICIDES A BASE DE PYRIMIDINE

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EP-A- 0 382 375

- **Chemical Abstracts, vol. 112, 1990, (Columbus,
Ohio, US), P. VAINILAVICHUS et al.:
"(6-Phenoxy-4-pyrimidinylthio)acetates:
synthesis and biological activity", see page 60,
column 1, abstract no. 612g, & KHIM.-FARM. ZH.
1989, 23(6), 705-7, see abstract**

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Description

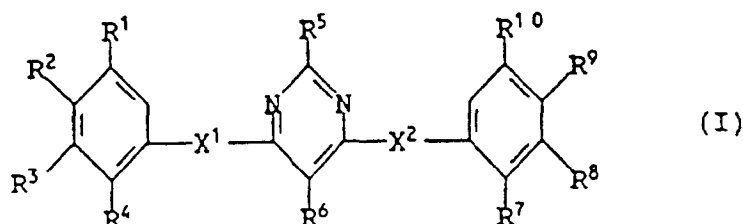
[0001] The present invention relates to substituted pyrimidine compounds, their preparation and use as pesticides.

[0002] Dutch Patent Specification No. 6814057 discloses a wide range of substituted pyrimidines and their use as fungicides.

[0003] J. Indian Chem. Soc., 52(8), 1975, 774-775, and 53(9), 1976, 913-914 discloses a number of 2-amino-4,6-bis aryloxy and arylimino pyrimidines and suggests that they may have useful biological properties.

[0004] It has now been found that a group of substituted pyrimidines generically described in NL-6814057 but not specifically disclosed therein have acaricidal activity which is significantly greater than that of 2-amino-substituted analogues.

[0005] The present invention provides a compound of the general formula



in which

X¹ and X² are the same and each represents an oxygen atom; a group S(O)_n in which n is 0, 1 or 2; or a group CO, CH₂ or NR in which R represents a hydrogen atom or a C₁₋₁₂ alkyl group;

R¹ and R¹⁰ are the same or different and each represents a hydrogen atom or a halogen atom;

R² and R⁹ are the same or different and each represents a hydrogen atom, a halogen atom or a cyano, nitro, C₁₋₁₂ alkyl, haloC₁₋₆alkyl, C₁₋₆alkoxy, C₁₋₆alkylthio, amino, mono- or di-C₁₋₆alkylamino, C₁₋₆alkoxycarbonyl, haloC₁₋₆alkoxycarbonyl or C₁₋₆alkoxycarbonyl group;

R³ and R⁸ are the same or different and each represents a hydrogen atom, a fluorine or chlorine atom, or a C₁₋₁₂ alkyl, haloC₁₋₆alkyl, haloC₁₋₆alkoxy, haloC₁₋₆alkylthio, haloC₂₋₄alkenyl, haloC₁₋₆alkoxycarbonyl, haloC₁₋₆alkylsulphanyl, haloC₁₋₆alkylsulphonyl, nitro or cyano group;

R⁴ and R⁷ are the same or different and each represents a hydrogen atom, a halogen atom or a C₁₋₁₂ alkyl or C₁₋₆ alkoxy group;

R⁵ represents a hydrogen atom, a halogen atom, or a cyano, C₁₋₁₂alkyl, haloC₁₋₆alkyl, C₁₋₆alkoxy, C₁₋₆alkylthio, C₁₋₆alkylsulphanyl or phenyl group;

and

R⁶ represents a hydrogen atom or, when R⁵ is hydrogen, a C₁₋₁₂alkyl group;

provided that either each of the two phenyl rings is unsubstituted or at least one of R³ and R⁸ is other than hydrogen.

[0006] To maintain activity the phenyl rings of formula I must be either unsubstituted or at least one must be 3-substituted.

[0007] An alkyl group, unless otherwise specified, is suitably a straight chain or branched chain group containing up to 12 carbon atoms, for example up to 8 carbon atoms. Preferably an alkyl group contains up to 6 carbon atoms. Especially preferred alkyl groups are methyl, ethyl and butyl. Any alkyl moiety which forms part of another group, for example the alkyl of a haloalkyl group or each alkyl of an alkoxyalkyl group, suitably has up to 6 carbon atoms, preferably up to 4 carbon atoms. Preferred alkyl moieties are methyl and ethyl.

[0008] Halogen is fluorine, chlorine, bromine or iodine. Haloalkyl and haloalkoxy are especially trifluoromethyl, pentafluoroethyl, and trifluoromethoxy.

[0009] Preferably each of X¹ and X² represents an oxygen atom, a sulphur atom or an NH group; especially each of X¹ and X² represents an oxygen atom

[0010] R¹ and R¹⁰ are preferably the same and each represents a hydrogen or fluorine atom, especially a hydrogen atom.

[0011] R² and R⁹ are preferably the same or different and each represents a hydrogen atom, a halogen atom, especially fluorine, chlorine or bromine, a nitro, alkyl, especially butyl, or cyano group.

[0012] R^3 and R^8 are preferably the same or different, each representing a hydrogen, fluorine or chlorine atom, or a nitro, C_{1-4} alkyl, halo- C_{1-4} alkyl, halo- C_{1-4} alkoxy, halo- C_{2-4} alkenyl or (C_{1-4} alkoxy)carbonyl group. In especially preferred compounds each of R^3 and R^8 represents a hydrogen or chlorine atom or a trifluoromethyl, trifluoromethoxy, pentafluoroethyl or difluoroethenyl group or one of R^3 and R^8 represents a trifluoromethyl group and the other represents a hydrogen, chlorine or fluorine atom or a methyl, butyl, nitro, cyano or methoxycarbonyl group.

[0013] R^4 and R^7 are preferably the same or different and each represents a hydrogen or halogen atom or a C_{1-4} alkyl group.

[0014] The pyrimidine ring, apart from the substituents at the 4- and the 6-positions, may carry one other substituent. R^5 , in the 2-position, preferably represents a hydrogen or halogen atom or a halo C_{1-4} alkyl, C_{1-4} alkylthio, C_{1-4} alkylsulphinyl or phenyl group, especially a hydrogen, fluorine, chlorine or bromine atom or a methylthio or ethylthio group. R^6 , in the 5-position, preferably represents a hydrogen atom or, when R^5 is hydrogen, a methyl group; R^6 is, however, especially hydrogen.

[0015] The compounds of formula I may be prepared by appropriate adaptation of conventional methods for obtaining disubstituted pyrimidines.

[0016] Conveniently, the compounds of formula I may be prepared by coupling appropriately substituted phenol(s), thiophenol(s) or aniline(s) and 4,6-dihalopyrimidines in basic conditions, optionally using a solvent, at ambient or, if necessary, at elevated temperatures, for example in the range of from 50 to 150°C. Desirably the reaction is carried out under nitrogen. Such procedures are well known and are described in, for example, J. Indian Chem. Soc. 52(8), 1975, 774-775, and 53(9), 1976, 913-914.

[0017] Naturally for the preparation of symmetrically substituted pyrimidine compounds of formula I, the reaction can be carried out in one step by using a molar ratio of pyrimidine to phenyl compound of at least 1:2. For unsymmetrical compounds, separate introduction of the two aryl substituents is required by a two-stage process.

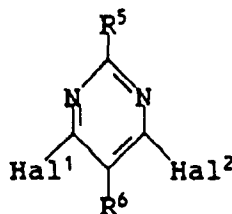
[0018] The basic conditions may be provided using an alkali metal salt, conveniently a sodium or potassium salt, e. g. an alkali metal hydride or carbonate, such as sodium hydride or potassium carbonate or other conventional bases such as n-butyllithium. The solvent, if used, may be any polar organic solvent and must be selected to be compatible with the base utilised in the reaction. Thus with potassium carbonate, dimethylformamide or dimethylsulphoxide are both suitable, and with sodium hydride, tetrahydrofuran may be used.

[0019] It is also possible to generate a 2- substituted 4,6-disubstituted pyrimidine from a corresponding compound with a different 2-substitution by standard procedures. Thus, for example, a 2-halo-4,6- disubstituted pyrimidine may be prepared from a corresponding 2-amino compound using an alkyl nitrite, for example *tert*-butylnitrite, and a suitable solvent, such as carbon tetrachloride; also a 2-hydroxy- 4,6-disubstituted pyrimidine may be converted into a 2-halo-analogue under the action of a phosphoryl halide, for example phosphoryl chloride or phosphoryl bromide, at an elevated temperature conveniently in the range of from 100 to the boiling temperature of the reaction medium; a reaction temperature of 130 to 150°C is very suitable for this type of reaction.

[0020] Furthermore, it is possible and, for some compounds of the invention, more convenient, to prepare certain compounds of formula I from other compounds of formula I by standard techniques. Thus, for example, the SO or SO₂ oxides of compounds in which X¹ and X² are each sulphur may be prepared by conventional oxidation techniques; the N-alkyl analogues of NH compounds may be prepared by standard alkylation procedures, e.g. using methyl iodide in triethylamine or with hydrogenation involving a palladium-carbon catalyst; and the 2-alkoxy compounds may be prepared from 2-chloro analogues using sodium alkoxide in methanol.

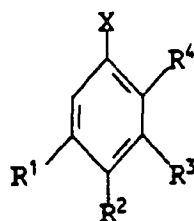
[0021] Therefore, the present invention further provides a process for the preparation of a compound of general formula I, which comprises

a) to prepare symmetrical compounds in which $R^1=R^{10}$, $R^2=R^9$, $R^3=R^8$ and $R^4=R^7$, reacting under basic conditions a 4,6-dihalopyrimidine of the general formula



(II)

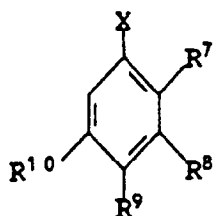
in which R^5 and R^6 are as defined above and each of Hal¹ and Hal², independently, represents a halogen atom, preferably chlorine or bromine, with a compound of the general formula



(III)

in which X represents a group CH_2Hal , COHal , OH , SH or NRH , Hal represents a halogen atom, suitably chlorine or bromine, and R, R^1 , R^2 , R^3 and R^4 are as defined above, in a molar ratio of at least 1:2;

b) to prepare unsymmetrical compounds in which R^1 , R^2 , R^3 and R^4 are not the same as R^{10} , R^9 , R^8 and R^7 respectively, reacting under basic conditions a compound of formula II with a compound of formula III in a molar ratio of 1:1 and then reacting the resulting product with a compound of the general formula

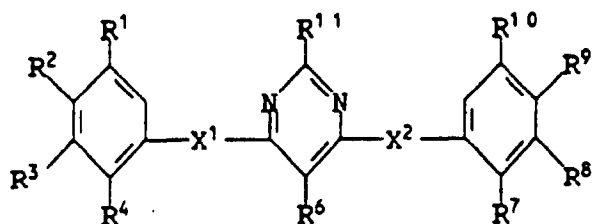


(IV)

in which X, R^7 , R^8 , R^9 and R^{10} are as defined above, also in a molar ratio of 1:1;

or

c) converting a compound of the general formula



(V)

in which X^1 , X^2 , R^1 , R^2 , R^3 , R^4 , R^6 , R^7 , R^8 , R^9 and R^{10} are as defined above, and R^{11} represents a group OH or NH_2 , into a compound of general formula I,

and, if desired or required, converting one compound of formula I into another compound of formula I.

[0022] The prepared compounds of formula I may, if desired, be isolated and purified using conventional techniques.

[0023] The compounds of formula II are either known or preparable by standard techniques, for example by the conversion of a corresponding pyrimidinol, (prepared following the Principal synthesis from the appropriate malonate and formamidine under reflux and in the presence of ethanol and sodium ethanolate) using a phosphoryl halide, e.g. chloride, in triethylamine at elevated temperature, for example at 100°C , as described in J. Org. Chem. 26, 1961, 4504.

[0024] Compounds of general formulae III and IV are either known or preparable by standard techniques, see for example J. Am. Chem. Soc. 73, 1951, 3470, which describes conditions suitable for the preparation of suitable phenols from the corresponding anilines using sodium nitrite and aqueous sulphuric acid at 0°C followed by steam distillation.

[0025] Compounds of general formula V, and alkyl derivatives thereof, with the exception of 2-amino-4,6-bisphenoxypyrimidine which is disclosed in J. Indian Chem. Soc. 53(9), 1976, 913-914, and 2-amino-4,6-bis(3-chlorophenylimino)pyrimidine which is disclosed in J. Indian Chem. Soc. 52(8), 1975, 774-775, are believed to be novel and also form part of the present invention. They may be prepared by methods analogous to that for the preparation of the compounds of formula I. The 2-hydroxy-4,6-dihalo-pyrimidine precursor required for the preparation of compounds of formula V in

which R¹¹ is hydroxy, may be prepared by procedures described in *Helv. Chim. Acta*, 72, 1989, 738, from 2,4,6-trihalo-*pyrimidine* reacted with dioxane in aqueous sodium hydroxide solution at ambient temperature. The other precursor compounds may also be prepared using standard literature procedures. The prime use of compounds of formula V is in the preparation of compounds of formula I, however one or two of the novel compounds of formula V unexpectedly possess pesticidal activity.

[0026] The compounds of the general formula I exhibit interesting and useful pesticidal, particularly acaricidal, activity and as such can be used to advantage to combat mites of the species *Tetranychus* and *Panonychus*. Moreover compounds of the present invention have been found to exhibit good activity against mite species which have developed resistance to existing commercial acaricides.

[0027] Certain compounds of the general formula I not only possess acaricidal activity but also exhibit useful activity against insect pests including whitefly and mosquito.

[0028] Furthermore, it has been found that compounds of the general formula I exhibit activity against animal ectoparasites, for example ticks on animals such as cattle, sheep, goats, pigs, dogs, horses, deer and cats.

[0029] The present invention therefore also provides a pesticidal composition comprising a carrier, preferably two carriers at least one of which is a surface-active agent, and, as active ingredient, a compound of general formula I. The invention additionally provides a method of combating pests, being primarily acarid pests, at a locus which comprises treating the locus with a compound or composition of the invention, and specifically provides the use as a pesticide, primarily as an acaricide, of a compound of general formula I. The dosage of active ingredient used may, for example, be from 5 to 500 ppm, preferably from 10 to 400 ppm, depending on the locus to be treated.

[0030] The present invention further provides a method of combating animal ectoparasites which comprises applying on to the skin or coat of an animal a compound of general formula I or a composition comprising such a compound as active ingredient.

[0031] A carrier in a composition according to the invention is any material with which the active ingredient is formulated to facilitate application to the locus to be treated, which may for example be a plant, seed or soil, or to facilitate storage, transport or handling. A carrier may be a solid or a liquid, including a material which is normally gaseous but which has been compressed to form a liquid, and any of the carriers normally used in formulating pesticidal compositions may be used. Preferably compositions according to the invention contain 0.5 to 95% by weight of active ingredient.

[0032] Suitable solid carriers include natural and synthetic clays and silicates, for example natural silicas such as diatomaceous earths; magnesium silicates, for example talcs; magnesium aluminium silicates, for example attapulgites and vermiculites; aluminium silicates, for example kaolinites, montmorillonites and micas; calcium carbonate; calcium sulphate; ammonium sulphate; synthetic hydrated silicon oxides and synthetic calcium or aluminium silicates; elements, for example carbon and sulphur; natural and synthetic resins, for example coumarone resins, polyvinyl chloride, and styrene polymers and copolymers; solid polychlorophenols; bitumen; waxes; and solid fertilisers, for example superphosphates.

[0033] Suitable liquid carriers include water; alcohols, for example isopropanol and glycols; ketones, for example acetone, methyl ethyl ketone, methyl isobutyl ketone and cyclohexanone; ethers; aromatic or aliphatic hydrocarbons, for example benzene, toluene and xylene; petroleum fractions, for example kerosine and light mineral oils; chlorinated hydrocarbons, for example carbon tetrachloride, perchloroethylene and trichloroethane. Mixtures of different liquids are often suitable.

[0034] Agricultural compositions are often formulated and transported in a concentrated form which is subsequently diluted by the user before application. The presence of small amounts of a carrier which is a surface-active agent facilitates this process of dilution. Thus preferably at least one carrier in a composition according to the invention is a surface-active agent. For example the composition may contain at least two carriers, at least one of which is a surface-active agent.

[0035] A surface-active agent may be an emulsifying agent, a dispersing agent or a wetting agent; it may be nonionic or ionic. Examples of suitable surface-active agents include the sodium or calcium salts of polyacrylic acids and lignin sulphonic acids; the condensation products of fatty acids or aliphatic amines or amides containing at least 12 carbon atoms in the molecule with ethylene oxide and/or propylene oxide; fatty acid esters of glycerol, sorbitol, sucrose or pentaerythritol; condensates of these with ethylene oxide and/or propylene oxide; condensation products of fatty alcohol or alkyl phenols, for example p-octylphenol or p-octylcresol, with ethylene oxide and/or propylene oxide; sulphates or sulphonates of these condensation products; alkali or alkaline earth metal salts, preferably sodium salts, of sulphuric or sulphonic acid esters containing at least 10 carbon atoms in the molecule, for example sodium lauryl sulphate, sodium secondary alkyl sulphates, sodium salts of sulphonated castor oil, and sodium alkylaryl sulphonates such as dodecylbenzene sulphonate; and polymers of ethylene oxide and copolymers of ethylene oxide and propylene oxide.

[0036] The compositions of the invention may for example be formulated as wettable powders, dusts, granules, solutions, emulsifiable concentrates, emulsions, suspension concentrates and aerosols. Wettable powders usually contain 25, 50 or 75% w of active ingredient and usually contain in addition to solid inert carrier, 3-10% w of a dispersing agent and, where necessary, 0-10% w of stabiliser(s) and/or other additives such as penetrants or stickers. Dusts are

usually formulated as a dust concentrate having a similar composition to that of a wettable powder but without a dispersant, and are diluted in the field with further solid carrier to give a composition usually containing ½-10% w of active ingredient. Granules are usually prepared to have a size between 10 and 100 BS mesh (1.676 - 0.152 mm), and may be manufactured by agglomeration or impregnation techniques. Generally, granules will contain ½-75% w active ingredient and 0-10% w of additives such as stabilisers, surfactants, slow release modifiers and binding agents. The so-called "dry flowable powders" consist of relatively small granules having a relatively high concentration of active ingredient. Emulsifiable concentrates usually contain, in addition to a solvent and, when necessary, co-solvent, 10-50% w/v active ingredient, 2-20% w/v emulsifiers and 0-20% w/v of other additives such as stabilisers, penetrants and corrosion inhibitors. Suspension concentrates are usually compounded so as to obtain a stable, non-sedimenting flowable product and usually contain 10-75% w active ingredient, 0.5-15% w of dispersing agents, 0.1-10% w of suspending agents such as protective colloids and thixotropic agents, 0-10% w of other additives such as defoamers, corrosion inhibitors, stabilisers, penetrants and stickers, and water or an organic liquid in which the active ingredient is substantially insoluble; certain organic solids or inorganic salts may be present dissolved in the formulation to assist in preventing sedimentation or as anti-freeze agents for water.

[0037] Aqueous dispersions and emulsions, for example compositions obtained by diluting a wettable powder or a concentrate according to the invention with water, also lie within the scope of the invention. The said emulsions may be of the water-in-oil or of the oil-in-water type, and may have a thick 'mayonnaise'-like consistency.

[0038] The composition of the invention may also contain other active ingredients, for example insecticides or fungicides, or, in appropriate circumstances, herbicides. The compounds of formula I may be found to be especially useful when applied in admixture with other insecticides and/or acaricides, e.g. organophosphates, pyrethroids, carbamates, acyl ureas and organotin compounds, for example the commercial products azinphos-methyl, chlorpyrifos, phosalone, fenpropathin, bifenthrin, pirimicarb, triazamate, diflubenzuron, flufenoxuron, teflubenzuron and fenbutatin oxide. Other mixture partners which, with the compounds of the invention may yield useful control, are amitraz, hexythiazox, pyridaben, and fenpyroximate.

[0039] The following Examples illustrate the invention. Examples 1 and 2 illustrate the preparation of starting materials of formulae III and II respectively; Examples 3 to 6 illustrate the preparation of compounds of formula I.

Example 1

Preparation of 4-bromo-3-trifluoromethylphenol

[0040] 4-Bromo-3-trifluoromethylaniline (48g, 0.2 mol) was treated with water (300 ml) and concentrated H₂SO₄ (36 ml) at 60°C for 1 hour. The resulting suspension was cooled in an ice bath and treated with sodium nitrite (16g, 0.23 mol) in water (30 ml) maintaining the temperature of the reaction mixture below 10°C. The resulting solution was stirred at 0°C for 1 hour, and then added portionwise, over 1 hour, to a 25% H₂SO₄ aqueous solution (160 ml) whilst steam distilling. After collecting approximately 1 litre of distillate, the aqueous distillate was extracted with ether and the organic solution dried using MgSO₄, filtered and concentrated. The product, 4-bromo-3-trifluoromethylphenol, was obtained by distillation under reduced pressure. Yield 18.0 g (37%); boiling point 68-71°C/133.3 Pa (1 mmHg)

Elemental Analysis (%)	Calculated	C 34.9	H 1.7
	Found	C 34.9	H 1.7

Example 2

Preparation of 4,6-dichloro-2-trifluoromethylpyrimidine

[0041] Sodium (13g, 0.57 mol) was dissolved in ethanol (500 ml) and diethyl malonate (84g, 0.53 mol) was added, followed by trifluoromethylformamidine (62g, 0.55 mol). The mixture was heated under reflux for 12 hours. On cooling, the mixture was concentrated under reduced pressure, and the product was taken up in water. On acidification with concentrated HCl, the product precipitated and was collected. Yield: 27.5g (28%).

[0042] The precipitate (5.0g, 0.028 mol) was suspended in triethylamine (20 ml) and was treated carefully with POCl₃ (20 ml). After the exotherm had subsided, the reaction mixture was heated at 100°C for 2 hours, and then cooled and poured onto ice. The product was extracted into diethyl ether, dried over Na₂SO₄ and concentrated under reduced pressure. The final product, 4,6-dichloro-2-trifluoromethylpyrimidine, was obtained by bulb-to-bulb distillation. Yield: 3.2g (52%); boiling point: 120°C/2666.4 Pa (20 mmHg).

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Example 3

Preparation of 4,6-bis(4-chloro-3-trifluoromethylphenoxy)pyrimidine

[0043] 4-Chloro-3-trifluoromethylphenol (10.0g, 0.051 mol) and 4,6-dichloropyrimidine (3.7g, 0.025 mol) were heated to 60°C in dimethylsulphoxide (75 ml) with potassium carbonate (10g) under nitrogen for 12 hours. The mixture was then poured into water and the product extracted into diethyl ether. The organic layer was dried using Na₂SO₄, filtered and concentrated. The product 4,6-bis(4-chloro-3-trifluoromethylphenoxy)pyrimidine, was obtained by column chromatography (eluting with 5:1, hexane ethyl acetate) and recrystallization (diethyl ether/hexane). Yield 11.0g (94%); melting point 111°C.

Elemental Analysis (%):				
Calculated	C 46.1	H 1.7	N 6.0	
Found	C 47.3	H 1.8	N 5.9	

Example 4

Preparation of 4,6-bis(3-trifluoromethylphenoxy)-2-bromo-pyrimidine

a) Preparation of 4,6-bis(3-trifluoromethylphenoxy)pyrimidin-2-one

[0044] Sodium hydroxide (20g, 0.5 mol) in water (160 ml) was added to a solution of 2,4,6-tri-chloropyrimidine (36.7g, 0.2 mol) in dioxane (600 ml). The mixture was stirred for 4 hours to give a thick white precipitate. The mixture was concentrated in vacuo and the residue recrystallized from boiling water. Yield: 18g (55%).

[0045] The residue, 4,6-dichloropyrimidin-2-one, (8.0g, 0.049 mol) and 3-trifluoromethylphenol (20g, 0.123 mol) were heated in dimethylformamide (250 ml) with potassium carbonate (16g) under nitrogen at 100°C for 12 hours. The mixture was then poured into water and the precipitate collected. The product, 4,6-bis(3-trifluoromethylphenoxy)pyrimidin-2-one, was obtained by recrystallization from methanol/water and column chromatography (eluting with 1:1, hexane:ethyl acetate). Yield: 2.5 g (12%).

Elemental Analysis (%):				
Calculated	C 44.6	H 1.7	N 5.8	
Found	C 45.7	H 2.1	N 5.7	

b) Preparation of 4,6-bis(3-trifluoromethylphenoxy)-2-bromopyrimidine

[0046] 4,6-Bis(3-trifluoromethylphenoxy)-pyrimidin-2-one (4.0g,

[0047] 0.0096 mol) and POBr₃ (100g) were heated at 140°C for 48 hours. The mixture was then poured onto a mixture of 2N NaOH (500 ml) and ice. The product was extracted into diethyl ether, dried using Na₂SO₄, filtered and concentrated under reduced pressure. The product was obtained by column chromatography (3:1, hexane: ethyl acetate) and recrystallization (ethyl acetate/hexane). Yield: 1.0g (22%); melting point 126-129°C.

Elemental Analysis (%):				
Calculated	C 45.1	H 1.9	N 5.9	
Found	C 45.7	H 2.1	N 6.1	

Example 5

Preparation of 4,6-bis(4-fluoro-3-trifluoromethyl-phenoxy)-2-chloro-pyrimidine

[0048] 2-Amino-4,6-bis(4-fluoro-3-trifluoromethylphenoxy)pyrimidine (3.0g, 6.7 mmol) was dissolved in carbon tetrachloride (75 ml) and the resulting solution was treated with t-butyl nitrite (1.2 ml, 13.4 mmol). The mixture was heated at 30°C for 48 hours and then poured into water. The product was extracted in dichloromethane, dried over sodium sulphate, filtered and concentrated under reduced pressure. The product, 4,6-bis(4-fluoro-3-trifluoromethylphenoxy)-2-chloro-pyrimidine, was obtained as an oil by column chromatography (eluting with 5:1, hexane:ethyl acetate). Yield

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0.3g (10%); mass 471 (M⁺+H); N.m.r: 7.3-7.5 (6H, m, aromatics), 6.35 (1H,s,H-5).

Elemental Analysis (%):				
Calculated	C 49.6	H 1.9	N 6.4	
Found	C 49.8	H 2.1	N 6.4	

Example 6

Preparation of 4-(4-cyanophenoxy)-6-(4-fluoro-3-trifluoromethylphenoxy)-pyrimidine

a) Preparation of 4-fluoro-6-(4-fluoro-3-trifluoromethylphenoxy)pyrimidine

[0049] 4,6-difluoropyrimidine (2.0 g, 0.017 mol) was placed in dimethylformamide (150 cm³) with potassium carbonate (2.5 g) and the temperature reduced to about -20°C. 4-fluoro-3-trifluoromethylphenol (2.9 g in 25 cm³ of dimethylformamide) was then added dropwise over 2 hours. The mixture was then left to stir for 4 hours between -30 and -20°C. After this time gas chromatography showed the reaction to be incomplete, so the mixture was left in the freezer overnight to prevent it from reaching room temperature. The mixture was then left to stir for a further 5 hours at -20°C after which time gas chromatography showed no further reaction. The mixture was then poured into water, the resultant solid filtered and recrystallised from cyclohexane. Yield 0.9 g (21%) :

Calculated	C 47.8	H 1.8	N 10.1
Found	C 48.0	H 2.2	N 10.1

A further 0.5 g of product was recovered from the recrystallisation filtrate, to give a total yield of 31%.

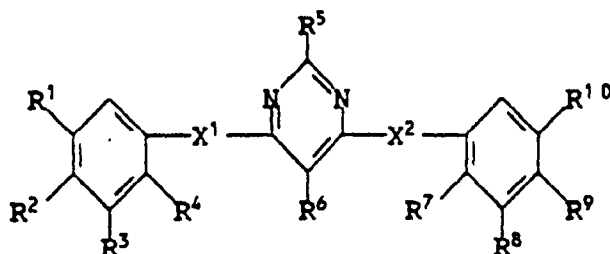
b) Preparation of 4-(4-cyanophenoxy)-6-(4-fluoro-3-trifluoromethylphenoxy)-pyrimidine

[0050] 4-fluoro-6-(4-fluoro-3-trifluoromethylphenoxy)-pyrimidine (0.9 g, 3.3 mmol) was placed in dimethylformamide (100 cm³) with potassium carbonate (0.6 g) and the temperature reduced to 0°C. 4-cyanophenol (0.37 g in 20 cm³ dimethylformamide) was then added dropwise and the mixture left to stir whilst maintaining the temperature at <5°C for 6 hours. After this time gas chromatography showed little or no reaction so a further 0.2 equivalent of 4-cyanophenol was added, and the mixture left to stir overnight with the temperature reaching room temperature. After this time, gas chromatography and thin layer chromatography showed the reaction to be complete, so the mixture was poured into 100 cm³ of water and the resultant solid filtered and recrystallised from cyclohexane. Yield 0.93 g (75%); melting point 131-132°C.

Calculated	C 57.6	H 2.4	N 11.2
Found	C 57.5	H 2.6	N 11.2

Examples 7 to 63

[0051] By methods analogous to those of Examples 3 to 6, further compounds of formula I were prepared. Details are given in Table I below, with reference to the following formula:



R⁴ and R⁷ are each hydrogen.

TABLE 1

Ex. No.	R ¹	R ²	R ³	X ¹	R ⁵	R ⁶	X ²	R ⁸	R ⁹	R ¹⁰	Melting Point (°C)	Elemental Analysis (% Calc./Found) C H N
7	H	H	CF ₃	O	H	H	O	CF ₃	H	H	72.0	54.0 2.5 7.0 54.2 2.6 7.4
8	H	H	CF ₃	S	H	H	S	CF ₃	H	H	157.0	50.0 2.3 6.5 50.4 2.2 7.0
9	H	H	CF ₃	NH	H	H	NH	CF ₃	H	H	195.0	54.3 3.0 14.1 54.1 3.5 14.4
10	H	H	CF ₃	O	H	CH ₃	O	CF ₃	H	H	82.3	55.1 2.9 6.8 54.9 3.1 6.7
11	H	H	OCF ₃	O	H	H	O	OCF ₃	H	H	oil	50.0 2.3 6.5 51.0 2.6 6.6
12	H	H	CF ₃	O	SCH ₃	H	O	CF ₃	H	H	86.4	51.1 2.7 6.3 50.9 2.9 6.2
13	H	H	CF ₃	O	CH ₃	H	O	CF ₃	H	H	104.9	55.1 2.9 6.8 56.1 3.0 6.8
14	H	H	CF ₃	O	SC ₂ H ₅	H	O	CF ₃	H	H	48.7	52.2 3.0 6.1 52.1 3.2 6.1

TABLE I (continued)

Ex. No.	R ¹	R ²	R ³	X ¹	R ⁵	R ⁶	X ²	R ⁸	R ⁹	R ¹⁰	Melting Point (°C)	Elemental Analysis (% Calc./Found)		
												C	H	N
15	H	H	CF ₃	O	SOC ₂ H ₅	H	O	CF ₃	H	H	82.9	50.4	2.9	5.9
												50.4	3.2	5.8
16	H	H	Cl	O	H	H	O	Cl	H	H	oil	57.9	3.0	8.4
												55.3	3.7	5.1
17	H	H	CF ₃	O	H	H	O	H	H	H	89.0	61.4	3.3	8.4
												61.1	3.5	8.1
18	H	NO ₂	CF ₃	O	H	H	O	CF ₃	NO ₂	H	136.0	44.1	1.6	11.4
												45.0	2.0	11.3
19	H	H	CF ₃	O	H	H	O	F	H	H	47.0	58.3	2.9	8.0
												58.8	3.1	8.1
20	H	H	CF ₃	O	H	H	O	Cl	H	H	78.0	55.7	2.7	7.6
												55.4	2.8	7.7
21	H	H	CF ₃	O	H	H	O	CH ₃	H	H	61.0	62.4	3.8	8.1
												62.3	4.0	7.9
22	H	H	C ₂ F ₅	O	H	H	O	C ₂ F ₅	H	H	oil	48.0	2.0	5.6
												48.4	2.3	5.7

TABLE 1 (continued)

Ex. No.	R ¹	R ²	R ³	X ¹	R ⁵	R ⁶	X ²	R ⁸	R ⁹	R ¹⁰	Melting Point (°C)	Elemental Analysis (% Calc./Found)		
												C	H	N
23	F	H	CF ₃	O	H	H	O	CF ₃	H	F	70.0	49.5 51.5	1.8 1.9	6.4 6.3
24	H	H	CF ₃	O	Cl	H	O	CF ₃	H	H	108-111	49.7 50.1	2.1 2.1	6.4 6.5
25	H	Br	CF ₃	O	H	H	O	CF ₃	Br	H	119.3- 121.4	38.7 38.9	1.5 1.8	5.0 5.1
26	H	Cl	CF ₃	O	Cl	H	O	CF ₃	Cl	H	135- 136	42.9 43.4	1.4 1.6	5.6 5.5
27	H	F	CF ₃	O	H	H	O	CF ₃	F	H	200°C* /0.6mmHg	49.6 49.9	1.8 2.4	6.4 5.6
28	H	Cl	CF ₃	O	SCH ₃	H	O	CF ₃	Cl	H	110.4- 110.5	44.4 44.7	2.0 2.3	5.3 5.4
29	H	CN	CF ₃	O	H	H	O	CF ₃	CN	H	132.0- 138.0	53.3 53.3	1.8 2.4	12.4 12.4
30	H	H	H	O	H	H	O	H	H	H	109.5	72.7 73.6	4.5 4.8	10.6 11.0

* boiling point

TABLE I (continued)

Ex. No.	R ¹	R ²	R ³	X ¹	R ⁵	R ⁶	X ²	R ⁸	R ⁹	R ¹⁰	Melting Point (°C)	Elemental Analysis (% Calc./Found)		
												C	H	N
31	H	H	H	O	SCH ₃	H	O	H	H	H	109.6	65.8	4.5	9.0
													4.5	8.9
32	H	Cl	CF ₃	O	CF ₃	H	O	CF ₃	Cl	H	116-120	42.5	1.3	5.2
												43.5	2.2	4.9
33	H	Cl	CF ₃	O	F	H	O	CF ₃	Cl	H	126.6-127.1	44.4	1.6	5.8
												44.3	1.9	5.7
34	H	H	CF ₃	O	C ₆ H ₅	H	O	CF ₃	H	H	84.0-85.0	60.5	3.0	5.9
												60.4	3.3	5.7
35	H	NO ₂	H	O	H	H	O	CF ₃	H	H	126.0-127.0	54.2	2.7	11.2
												54.2	2.7	12.0
36	H	F	H	O	H	H	O	CF ₃	H	H	117.9-118.4	59.0	2.9	8.0
												60.5	2.9	8.7
37	H	F	CF ₃	O	SCH ₃	H	O	CF ₃	F	H	97.0-98.0	47.3	2.1	5.8
												48.0	2.6	6.0
38	H	Cl	CF ₃	O	F	H	O	CF ₃	Cl	H	129.0-130.0	44.4	1.4	5.7
												44.3	1.7	5.7

TABLE 1 (continued)

Ex No.	R ¹	R ²	R ³	X ¹	R ⁵	R ⁶	X ²	R ⁸	R ⁹	R ¹⁰	Melting Point (°C)	Elemental Analysis (% Calc./Found)		
												C	H	N
39	H	F	H	O	H	H	O	CF ₃	F	H	122.0- 123.0	56.0 57.3	2.2 2.9	7.6 7.7
40	H	H	CF ₃	O	H	H	O	H	CN	H	196.0- 197.0	60.6 59.4	2.8 3.0	11.7 10.6
41	H	CN	CF ₃	O	SCH ₃	H	O	CF ₃	CN	H	148.0- 149.0	50.3 51.1	2.4 2.4	11.3 11.2
42	H	F	CF ₃	O	C ₆ H ₅	H	O	CF ₃	F	H	78.0- 79.0	56.3 56.0	2.4 2.5	5.5 5.3
43	H	H	C ₂ F ₅	O	SCH ₃	H	O	C ₂ F ₅	H	H	61.0- 62.0	46.1 46.1	2.2 2.4	5.1 5.1
44	H	H	CH-CF ₂	O	H	H	O	CH-CF ₂	H	H	oil	-	-	-
45	H	F	CF ₃	O	H	H	O	CF ₃	H	H	57.0- 58.0	51.8 51.8	2.4 2.4	6.7 6.7
46	H	F	CF ₃	O	H	H	O	H	NO ₂	H	oil	51.7 51.5	2.3 2.5	10.6 10.5
47	H	Br	CF ₃	O	Cl	H	O	CF ₃	Br	H	192.0- 193.0	36.5 36.3	1.2 1.5	4.7 4.7

TABLE I (continued)

Ex. No.	R ¹	R ²	R ³	X ¹	R ⁵	R ⁶	X ²	R ⁸	R ⁹	R ¹⁰	Melting Point (°C)	Elemental Analysis (% Calc./Found) C H N
48	H	F	CF ₃	O	H	H	O	H	Cl	H	oil	53.1 2.3 7.3 53.5 2.6 7.3
49	H	F	CF ₃	O	H	H	O	Cl	H	H	61.9- 62.7	53.1 2.3 7.3 53.3 2.6 7.2
50	H	F	CF ₃	O	H	H	O	F	H	H	oil	55.4 2.4 7.6 55.6 2.6 7.6
51	H	F	CF ₃	O	H	H	O	CH ₃	H	H	oil	59.3 3.3 7.3 59.6 3.4 7.7
52	H	H	C ₂ F ₅	O	Cl	H	O	C ₂ F ₅	H	H	85.0- 87.0	44.9 1.7 5.2 44.6 2.1 5.3
53	H	F	CF ₃	O	H	H	O	CF ₃	Cl	H	68.0	47.7 1.8 6.2 47.4 2.0 6.1
54	H	F	CF ₃	O	H	H	O	C(CH ₃) ₃	H	H	69.7- 71.6	62.1 4.4 6.9 62.9 4.8 6.9
55	H	F	CF ₃	O	H	H	O	H	C(CH ₃) ₃	H	104.6- 105.1	62.1 4.4 6.9 62.4 4.8 6.9

TABLE 1 (continued)

Ex. No.	R ¹	R ²	R ³	X ¹	R ⁵	R ⁶	X ²	R ⁸	R ⁹	R ¹⁰	Melting Point (°C)	Elemental Analysis (% Calc./Found) C H N
56	H	F	CF ₃	O	H	H	O	NO ₂	H	H	103.0-103.2	51.6 2.3 10.6 51.4 2.5 10.6
57	H	H	CF ₃	O	H	H	O	CF ₃	Cl	H	oil	-
58	H	F	CF ₃	O	H	H	O	CN	H	H	80.9-81.3	57.6 2.4 11.2 57.1 2.7 11.2
59	H	F	CF ₃	O	H	H	O	CO ₂ CH ₃	H	H	71.0-72.0	55.9 3.0 6.9 55.8 3.2 6.9
60	H	F	CF ₃	O	SCH ₃	H	O	CF ₃	H	H	86.5-87.2	49.0 2.4 6.0 48.7 2.7 6.3
61	H	Br	CF ₃	O	H	H	O	CF ₃	H	H	oil	45.1 2.0 5.8 45.0 2.4 5.9
62	H	Br	CF ₃	O	H	H	O	CF ₃	F	H	oil	43.5 1.6 5.6 43.9 2.0 5.6
63	H	F	CF ₃	O	Cl	H	O	CF ₃	H	H	105.3-105.5	47.8 1.8 6.2 46.7 2.1 6.2

Examples 64 to 67

[0052] By methods analogous to those of Examples 3 to 6, further compounds of formula I were prepared. Details are given in Table II below, with reference to the following formula:

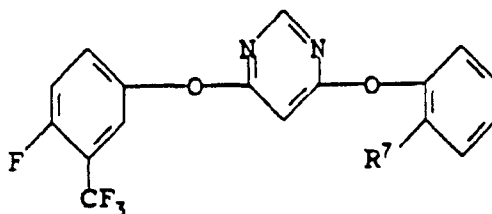


Table II

Ex No.	R ⁷	Melting Point (°C)	Elemental Analysis (% Calc./Found)		
			C	H	N
64	Cl	oil	53.1	2.3	7.3
			53.2	2.9	6.9
65	F	73.0-73.1	55.4	2.4	7.6
			55.7	2.7	7.5
66	C(CH ₃) ₃	oil	52.1	4.4	6.9
			52.7	4.8	6.6
67	CH ₃	oil	59.3	3.3	7.7
			59.7	3.3	7.4

Example 68

Acaricidal Activity

[0053] The acaricidal activity of the compounds of the invention was determined in the following tests employing the glasshouse red spider mite, *Tetranychus urticae* (T.u.).

[0054] In each test solutions or suspensions of test compound were made up over a range of concentrations in water (initially 0.1%w) containing 10%w acetone and 0.025%w "TRITON X-100" (trade mark) surface active agent (the condensation product of ethylene oxide with an alkyl phenol). These solutions were sprayed at a rate equivalent to 340 litres per hectare ($3.4 \times 10^{-5} \text{m}^3/\text{m}^2$) onto petri dishes containing either test species per se or diet onto which test species were subsequently introduced, as indicated. The tests were all conducted under normal insectary conditions ($23^\circ\text{C} \pm 2^\circ\text{C}$, fluctuating humidity and 16 hours day length light).

[0055] The results of testing at the initial test concentrations were graded:

Grade A represents at least 70% mortality of the pest

Grade B represents from 40% to 69% mortality.

[0056] For compounds achieving Grade A at initial test concentration, mortality assessments were made as indicated below, in terms of percentage mortality figures. In each test a LC_{50} (the dosage of active material required to kill half of the test species) for the compound was calculated from the mortality figures and compared with the corresponding LC_{50} for a standard insecticide (either ethyl parathion or chlorfenson, as indicated) in the same test. The results are expressed as toxicity indices thus:

$$\text{toxicity index} = \frac{\text{LC}_{50} \text{ (standard insecticide)}}{\text{LC}_{50} \text{ (test compound)}} \times 100$$

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a) Acaricidal activity - mite adults Tu

[0057] Acaricidal activity was assessed using adult glasshouse red spider mites, Tetranychus urticae (T.u.), 7-10 days after hatching, by the following procedure:

[0058] 2 cm diameter discs cut from the leaves of French bean plants were placed on filter paper kept moist by a cotton wool wick dipped into water. Prior to testing, each leaf disc was infested with 10 adult mites. The mites and discs were then sprayed with solutions of the test compound made up as above, at a rate equivalent to 340 litres per hectare ($3.4 \times 10^5 \text{ m}^3/\text{m}^2$). The mites were held under the normal insectary conditions. The numbers of dead and moribund adults were assessed after 48 hours and the percentage mortality calculated.

b) Acaricidal activity - ovicide TuOA

[0059] Acaricidal activity was assessed employing eggs of the glasshouse red spider mite, Tetranychus urticae (T.u.), less than 24 hours old, by the following procedure.

[0060] 2 cm diameter leaf discs cut from the leaves of French bean plants were placed on filter paper, kept moist by a cotton wool wick dipped into water.

[0061] On the day before spraying, each leaf disc was infested with 10 female adult mites. On the day of the test, the adults were removed, leaving the eggs laid overnight on the discs. The leaf discs were then sprayed with solutions of test compound made up as above, at a rate equivalent to 340 litres per hectare ($3.4 \times 10^5 \text{ m}^3/\text{m}^2$).

[0062] Throughout the test, the eggs were held under the normal insectary conditions. After 7-10 days, the numbers of hatched nymphs and unhatched eggs were assessed and the percentage mortality calculated.

[0063] The LC_{50} (the dosage of active material required to kill half of the test species) for each test compound was calculated from the mortality figure and compared with the corresponding LC_{50} for a standard insecticide in the same test. For Tu ethyl parathion was used as the standard compound; for TuOA chlorfenson was used as the standard.

[0064] The results are given in Table III below.

Table III

Acaricidal Activity			
Compound of Example No.	Toxicity Index		
	Tu	Tu OA	
3	320	720	
4	67		
5	400	1400	
6	12	140	
7	75	94	
8	5		
9	<4	<20	
10		B	
11	<3		
12	28	66	
13	5		
14	23	63	
15		<18	
16	5	20	
17		41	
18	70	34	
19	10		
20	27	35	
21	4	12	
22	100	41	
23	4		
24	170	<16	
25	98	180	

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Table III (continued)

Acaricidal Activity			
Compound of Example No.	Toxicity Index		
	Tu	Tu OA	
26	150	180	
27	360	2200	
28	18	B	
29	62	250	
30		<23	
31		<12	
32	22	73	
33	60	200	
34	11	<19	
35	25	<16	
36	21	60	
37	190	730	
38	100		
39	42	390	
40	35	87	
41	10	94	
42	36	20	
43		57	
44		36	
45	110	640	
46	73	100	
47	18		
48	21	160	
49	77	190	
50	24	170	
51	38	65	
52	100	130	
53	760	1500	
54	29	81	
55	11	62	
56	128	700	
57	94	870	
58	72	650	
59	6	<4	
60	65	A	
61	57	A	
62	130	A	
63	120	760	
64	2	<27	
65	10	50	
66	11	57	
67	19	100	

Example 69Comparison Tests

[0065] The acaricidal activity of the prior art compound 2-amino-4,6-bisphenoxy-pyrimidine and of the 2-amino analogues of the substituted pyrimidines of Examples 3 and 7 was determined following the procedures of Example 68 above. The results are given in Table IV below, and for ease of comparison the data for the compounds of Examples 3 and 7 are also included in the Table.

Table IV

Compound	Toxicity Index	
	Tu	TuOA
Example 3	320	720
Example 7	75	94
Comparison A	C	C
Comparison B	C	C
Comparison C	C	C

[0066] Comparison A is the 2-amino analogue of the compound of Example 3, and Comparison B is the 2-amino analogue of the compound of Example 7. Comparison C is 2-amino-4,6-bisphenoxy-pyrimidine. Grade C represents less than 40% mortality of the pest, whereas the Toxicity Index is only estimated when Grade A activity (i.e. at least 70% mortality) is achieved.

[0067] It can clearly be seen that the compounds of the present invention have a significantly greater acaricidal activity than the direct 2-amino analogues.

Example 70Insecticidal Activity

[0068] Insecticidal activity of compounds of general formula I was assessed against the following pest:-

Trialeurodes vaporariorum (greenhouse whitefly)(T.v.)

[0069] The test method employed appears below. In each test, solutions or suspensions of test compound were made up and sprayed as described above in Example 68.

[0070] French bean plants (Phaseolus vulgaris) with two fully expanded leaves were placed in a breeding culture of T.vaporariorum, also on French bean plants, which were then disturbed to ensure resettlement on the introduced plants. During the subsequent 24 hour period, eggs were deposited and kept at 27°C, with 14 hours photoperiod. All adult whiteflies were then carefully removed, leaving egg samples of a known age. After eight days the majority of eggs had hatched. Leaf discs containing the newly hatched nymphs were then cut from the leaves and transferred to moist filter paper. The discs were examined under a low-powered microscope to determine the exact number of 1st instar nymphs per disc and to remove any unhatched eggs. On average, 70-100 nymphs were found per disc. The discs were transferred into Petri dishes and sprayed with test solutions as described above. After 6 days percentage mortalities were assessed.

[0071] The LC₅₀ for each test compound was calculated as described above in Example 68. Ethyl parathion was used as the standard compound. The results are given in Table V below.

Table V

Insecticidal Activity	
Compound of Example No.	Toxicity Index
	T.v.
5	320
22	180
27	95
36	<20
37	190

Table V (continued)

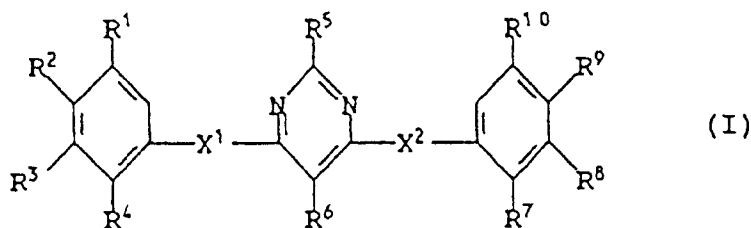
Insecticidal Activity	
Compound of Example No.	Toxicity Index
	T.v.
39	95
45	370
48	17
49	27

Example 71Ectoparasiticial Activity

[0072] In tests on tick larvae, *Boophilus decoloratus*, a concentration range of 1-25 ppm of the compound of Example 27 was used. Dead larvae were detected after 24 hours with all the concentrations, with the highest mortality occurring with the 25 ppm concentration. After 40 hours, no live larvae were observed with the 25 ppm concentration.

Claims

1. A compound of the general formula



in which

X¹ and X² are the same and each represents an oxygen atom; a group S(O)_n in which n is 0, 1 or 2; or a group CO, CH₂ or NR in which R represents a hydrogen atom or a C₁₋₁₂ alkyl group;

R¹ and R¹⁰ are the same or different and each represents a hydrogen atom or a halogen atom;
 R² and R⁹ are the same or different and each represents a hydrogen atom, a halogen atom or a cyano, nitro, C₁₋₁₂ alkyl group; haloC₁₋₆alkyl, C₁₋₆alkoxy, C₁₋₆alkylthio, amino, mono- or di-C₁₋₆alkylamino, C₁₋₆alkoxyC₁₋₆alkyl, haloC₁₋₆alkoxyC₁₋₆alkyl or C₁₋₆alkoxycarbonyl group;

R³ and R⁸ are the same or different and each represents a hydrogen atom, a fluorine or chlorine atom, or a C₁₋₁₂ alkyl, haloC₁₋₆alkyl, haloC₁₋₆alkoxy, haloC₁₋₆alkylthio, haloC₂₋₄alkenyl, haloC₁₋₆alkoxyC₁₋₆alkyl, C₁₋₆alkoxycarbonyl, haloC₁₋₆alkylsulphinyl, haloC₁₋₆alkylsulphonyl, nitro or cyano group;

R⁴ and R⁷ are the same or different and each represents a hydrogen atom, a halogen atom or a C₁₋₁₂ alkyl or C₁₋₆alkoxy group;

R⁵ represents a hydrogen atom, a halogen atom, or a cyano, C₁₋₁₂alkyl, haloC₁₋₆alkyl, C₁₋₆alkoxy, C₁₋₆alkylthio, C₁₋₆alkylsulphinyl or phenyl group;

and

R⁶ represents a hydrogen atom or, when R⁵ is hydrogen, a C₁₋₁₂alkyl group;

provided that either each of the two phenyl rings is unsubstituted or at least one of R³ and R⁸ is other than hydrogen.

2. A compound as claimed in claim 1, in which each of X^1 and X^2 represents an oxygen atom, a sulphur atom or a group NH.

3. A compound as claimed in claim 2, in which each of X^1 and X^2 represents an oxygen atom.

4. A compound as claimed in any one of claims 1 to 3, in which

R^1 and R^{10} are the same and each represents a hydrogen or a fluorine atom;

R^2 and R^9 are the same or different and each represents a hydrogen atom, a halogen atom, a nitro, C_{1-6} alkyl or cyano group;

R^3 and R^8 are the same or different and each represents a hydrogen, fluorine or chlorine atom, or a nitro, C_{1-4} alkyl, halo C_{1-4} alkyl, halo C_{1-4} alkoxy, halo C_{2-4} alkenyl or (C_{1-4} alkoxy)carbonyl group;

R^4 and R^7 are the same or different and each represents a hydrogen or halogen atom or a C_{1-4} alkyl group;

R^5 represents a hydrogen atom, a halogen atom or halo C_{1-4} alkyl, a C_{1-4} alkylthio or C_{1-4} alkylsulphinyl or phenyl group; and

R^6 represents a hydrogen atom or, when R^5 is hydrogen, a methyl group.

5. A compound as claimed in any one of claims 1 to 3, in which each of R^1 and R^{10} represents a hydrogen atom;

each of R^2 and R^9 represents a hydrogen, fluorine, chlorine or bromine atom or a butyl, cyano or nitro group;

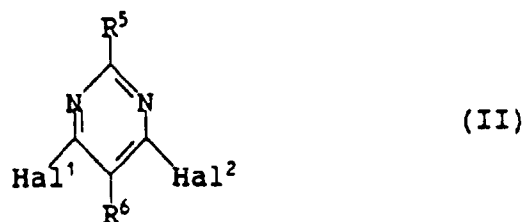
each of R^3 and R^8 represents a hydrogen or chlorine atom or a trifluoromethyl, trifluoromethoxy, pentafluoroethyl or difluoroethenyl group or one of R^3 and R^8 represents a trifluoromethyl group and the other represents a hydrogen, chlorine or fluorine atom or a methyl, butyl, nitro, cyano or methoxycarbonyl group;

R^5 represents a hydrogen, fluorine, chlorine or bromine atom, or a methyl, methylthio, ethylthio, ethylsulphinyl or phenyl group; and

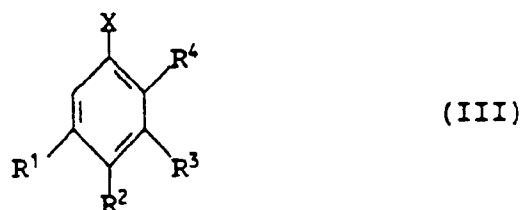
R^6 represents a hydrogen atom.

6. A process for the preparation of a compound of general formula I as claimed in claim 1, which comprises

a) to prepare symmetrical compounds in which $R^1=R^{10}$, $R^2=R^9$, $R^3=R^8$ and $R^4=R^7$, reacting under basic conditions a 4,6-dihalopyrimidine of the general formula



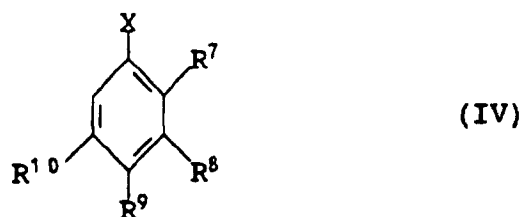
in which R^5 and R^6 are as defined in claim 1 and each of Hal^1 and Hal^2 , independently, represents a halogen atom, with a compound of the general formula



in which X represents a group CH_2Hal , $COHal$, OH, SH or NRH, Hal represents a halogen atom, and R, R^1 , R^2 , R^3 and R^4 are as defined in claim 1, in a molar ratio of at least 1:2;

b) to prepare unsymmetrical compounds in which R^1 , R^2 , R^3 and R^4 are not the same as R^{10} , R^9 , R^8 and R^7

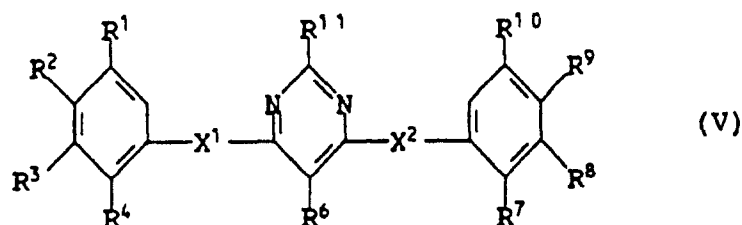
respectively, reacting under basic conditions a compound of formula II with a compound of formula III in a molar ratio of 1:1 and then reacting the resulting product with a compound of the general formula



in which X, R⁷, R⁸, R⁹ and R¹⁰ are as defined in claim 1, also in a molar ratio of 1:1;

or

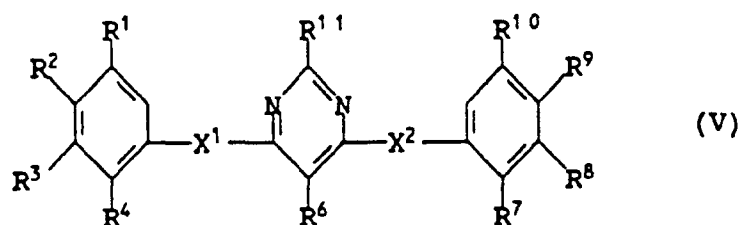
c) converting a compound of the general formula



in which X¹, X², R¹, R², R³, R⁴, R⁶, R⁷, R⁸, R⁹ and R¹⁰ are as defined in claim 1, and R¹¹ represents a group OH or NH₂, into a compound of general formula I,

and if desired or required, converting one compound of general formula I into another compound of general formula I.

7. A pesticidal composition comprising a carrier and, as active ingredient, a compound of formula I as claimed in any one of claims 1 to 5 herein.
8. A method of combating pests at a locus, which comprises treating the locus with a compound of formula I as claimed in any one of claims 1 to 5, or a composition as claimed in claim 7.
9. A method of combating animal ectoparasites which comprises applying on to the skin or coat of an animal a compound of formula I as claimed in any one of claims 1 to 5, or a composition as claimed in claim 7.
10. A compound of the general formula

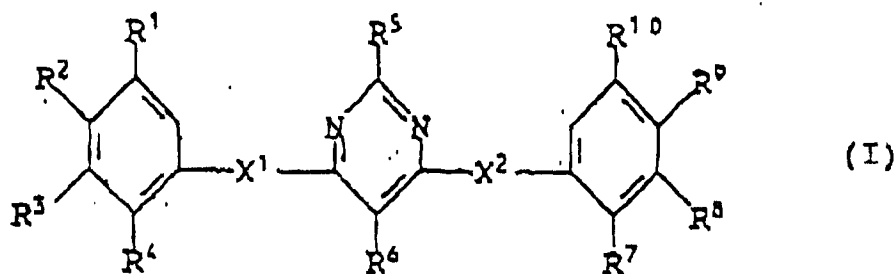


in which X¹, X², R¹, R², R³, R⁴, R⁶, R⁷, R⁸, R⁹, R¹⁰ and R¹¹ are as defined in claim 6, or a derivative thereof in which R¹¹ represents an alkoxy group or a mono- or di-alkylamino group, with the exception of 2-amino-4,6-bi-

sphenoxypyrimidine, and 2-amino-4,6-bis(3-chlorophenylimino)pyrimidine.

Patentansprüche

1. Verbindung der allgemeinen Formel



in der die Symbole die folgenden Bedeutungen haben:

X¹ und X² sind gleich und bedeuten jeweils ein Sauerstoffatom, eine Gruppe S(O)_n, in der n für 0, 1 oder 2 steht, oder eine Gruppe CO, CH₂ oder NR, in der R ein Wasserstoffatom oder eine C₁₋₁₂Alkylgruppe bedeutet,

R¹ und R¹⁰ sind gleich oder verschieden und bedeuten jeweils ein Wasserstoffatom oder ein Halogenatom,
R² und R⁹ sind gleich oder verschieden und bedeuten jeweils ein Wasserstoffatom, ein Halogenatom oder eine Cyan-, Nitro-, C₁₋₁₂Alkylgruppe, HaloC₁₋₆alkyl-, C₁₋₆Alkoxy-, C₁₋₆Alkylthio-, Amino-, Mono- oder Di-C₁₋₆alkyl-, C₁₋₆AlkoxyC₁₋₆alkylamino-, HaloC₁₋₆alkoxy-C₁₋₆alkyl- oder C₁₋₆Alkoxy-carbonylgruppe,

R³ und R⁸ sind gleich oder verschieden und bedeuten jeweils ein Wasserstoffatom, ein Fluor- oder Chloratom oder eine C₁₋₁₂Alkyl-, HaloC₁₋₆alkyl-, HaloC₁₋₆alkoxy, HaloC₁₋₆alkylthio-, HaloC₂₋₄alkenyl-, HaloC₁₋₆alkoxyC₁₋₆alkyl-, C₁₋₆Alkoxy-carbonyl-, HaloC₁₋₆alkylsulfinyl-, HaloC₁₋₆alkylsulfonyl-, Nitro- oder Cyangruppe,

R⁴ und R⁷ sind gleich oder verschieden und bedeuten jeweils ein Wasserstoffatom, ein Halogenatom oder eine C₁₋₁₂Alkyl- oder C₁₋₆Alkoxygruppe,

R⁵ bedeutet ein Wasserstoffatom, ein Halogenatom oder eine Cyan-, C₁₋₁₂Alkyl-, HaloC₁₋₆alkyl-, C₁₋₆Alkoxy-, C₁₋₆Alkylthio-, C₁₋₆Alkylsulfinyl- oder Phenylgruppe

und

R⁶ bedeutet ein Wasserstoffatom oder, wenn R⁵ Wasserstoff darstellt, eine C₁₋₁₂Alkylgruppe,

mit der Maßgabe, daß entweder keiner der beiden Phenylringe substituiert ist oder mindestens einer der Reste R³ und R⁸ eine andere Bedeutung als Wasserstoff hat.

2. Verbindung nach Anspruch 1, in der X¹ und X² jeweils ein Sauerstoffatom, ein Schwefelatom oder eine NH-Gruppe bedeuten.

3. Verbindung nach Anspruch 2, in der X¹ und X² jeweils ein Sauerstoffatom bedeuten.

4. Verbindung nach einem der Ansprüche 1 bis 3, in der R¹ und R¹⁰ gleich sind und jeweils ein Wasserstoff- oder ein Fluoratom bedeuten,

R² und R⁹ gleich oder verschieden sind und jeweils ein Wasserstoffatom, ein Halogenatom oder eine Nitro-, C₁₋₆Alkyl- oder Cyangruppe bedeuten,

R³ und R⁸ gleich oder verschieden sind und jeweils ein Wasserstoff-, Fluor- oder Chloratom oder eine Nitro-, C₁₋₄Alkyl-, HaloC₁₋₄alkyl-, HaloC₁₋₄alkoxy-, HaloC₂₋₄alkenyl- oder (C₁₋₄Alkoxy) carbonylgruppe bedeuten,
R⁴ und R⁷ gleich oder verschieden sind und jeweils ein Wasserstoff- oder Halogenatom oder eine C₁₋₄Alkylgruppe bedeuten,

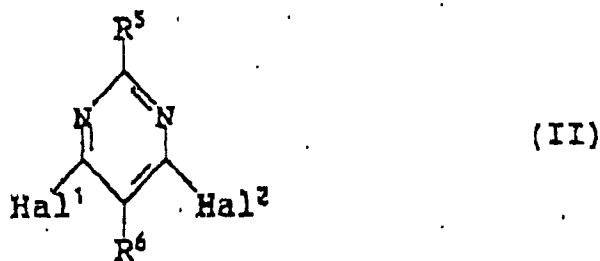
R⁵ ein Wasserstoffatom, ein Halogenatom oder eine HaloC₁₋₄alkyl-, C₁₋₄Alkylthio-, C₁₋₄Alkylsulfinyl- oder Phenylgruppe bedeutet, und
R⁶ ein Wasserstoffatom oder, falls R⁵ Wasserstoff darstellt, eine Methylgruppe bedeutet.

5. Verbindung nach einem der Ansprüche 1 bis 3, bei der

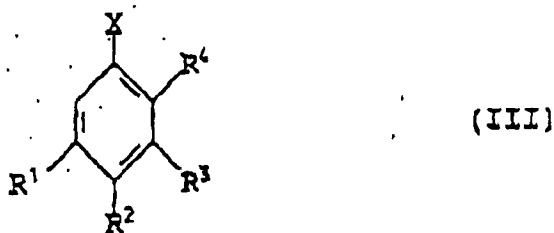
R¹ und R¹⁰ jeweils ein Wasserstoffatom bedeuten,
R² und R⁹ jeweils ein Wasserstoff-, Fluor-, Chlor- oder Bromatom oder eine Butyl-, Cyan- oder Nitrogruppe bedeuten,
R³ und R⁸ jeweils ein Wasserstoff- oder Chloratom oder eine Trifluormethyl-, Trifluormethoxy-, Pentafluorethyl- oder Difluorethenylgruppe bedeuten, oder einer der Reste R³ und R⁸ bedeutet eine Trifluormethylgruppe und der andere bedeutet ein Wasserstoff-, Chlor- oder Fluoratom oder eine Methyl-, Butyl-, Nitro-, Cyan- oder Methoxycarbonylgruppe,
R⁵ ein Wasserstoff-, Fluor-, Chlor- oder Bromatom oder eine Methyl-, Methylthio-, Ethylthio-, Ethylsulphinyl- oder Phenylgruppe bedeutet, und
R⁶ ein Wasserstoffatom bedeutet.

6. verfahren zur Herstellung einer Verbindung der allgemeinen Formel I nach Anspruch 1, **dadurch gekennzeichnet, daß** man

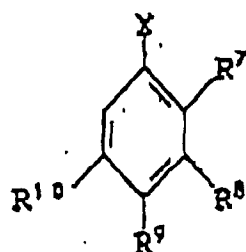
a) zur Herstellung von symmetrischen Verbindungen, in denen R¹=R¹⁰, R²=R⁹, R³=R⁸ und R⁴=R⁷ ist, ein 4,6-Dihalogenpyrimidin der allgemeinen Formel



in der R⁵ und R⁶ wie in Anspruch 1 definiert sind und Hal¹ und Hal² jeweils unabhängig ein Halogenatom darstellen, unter basischen Bedingungen mit einer Verbindung der allgemeinen Formel



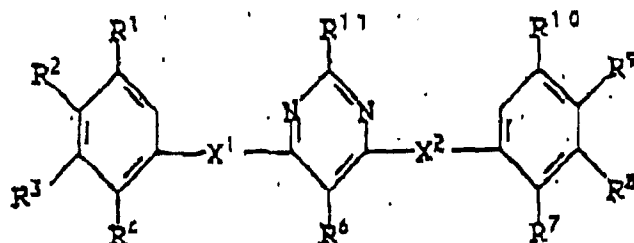
in der X eine Gruppe CH₂Hal, COHal, OH, SH oder NRH darstellt, Hal ein Halogenatom darstellt und R, R¹, R², R³ und R⁴ wie in Anspruch 1 definiert sind, in einem Molverhältnis von mindestens 1:2 umgesetzt,
b) zur Herstellung von asymmetrischen Verbindungen, in denen R¹, R², R³ und R⁴ nicht gleich R¹⁰, R⁹, R⁸ bzw. R⁷ sind, eine Verbindung der Formel II mit einer Verbindung der Formel III unter basischen Bedingungen in einem Molverhältnis von 1:1 umgesetzt und dann das entstandene Produkt mit einer Verbindung der allgemeinen Formel



(IV)

in der X, R⁷, R⁸, R⁹ und R¹⁰ wie in Anspruch 1 definiert sind, ebenfalls in einem Molverhältnis von 1:1 umgesetzt, oder

c) eine Verbindung der allgemeinen Formel



(V)

in der X¹, X², R¹, R², R³, R⁴, R⁶, R⁷, R⁸, R⁹ und R¹⁰ wie in Anspruch 1 definiert sind und R¹¹ eine OH- oder NH₂-Gruppe darstellt, in eine Verbindung der allgemeinen Formel I umwandelt,

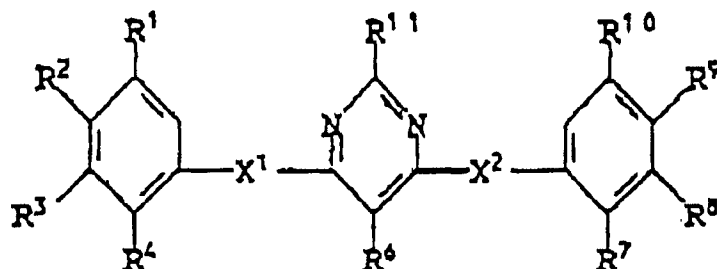
und, falls erwünscht oder erforderlich, eine Verbindung der allgemeinen Formel I in eine andere Verbindung der allgemeinen Formel I umwandelt.

7. Pestizides Mittel mit einem Träger sowie einer Verbindung der Formel I nach einem der vorliegenden Ansprüche 1 bis 5 als Wirkstoff.

8. Verfahren zur Kontrolle von Schädlingen an einem Ort, **dadurch gekennzeichnet, daß** man den Ort mit einer Verbindung der Formel I nach einem der Ansprüche 1 bis 5 oder einem Mittel nach Anspruch 7 behandelt.

9. Verfahren zur Bekämpfung von Ektoparasiten an Tieren, **dadurch gekennzeichnet, daß** man eine Verbindung der Formel I nach einem der Ansprüche 1 bis 5 oder ein Mittel nach Anspruch 7 auf die Haut oder das Fell eines Tiers aufträgt.

10. Verbindung der allgemeinen Formel



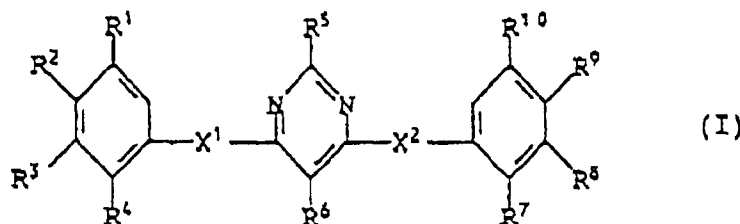
(V)

in der X¹, X², R¹, R², R³, R⁴, R⁶, R⁷, R⁸, R⁹, R¹⁰ und R¹¹ wie in Anspruch 6 definiert, oder ein Derivat davon sind,

in der R¹¹ eine Alkoxygruppe oder eine Mono- oder Dialkylaminogruppe bedeutet, mit Ausnahme von 2-Amino-4,6-bisphenoxyypyrimidin und 2-Amino-4,6-bis (3-chlorphenylimino)pyrimidin.

5 Revendications

1. Composé de formule générale



dans laquelle

X¹ et X² sont identiques ou différents et représentent chacun un atome d'oxygène ; un groupe S(O)_n dans lequel n est 0, 1 ou 2 ; ou un groupe CO, CH₂ ou NR dans lequel R représente un atome d'hydrogène ou un groupe alkyle en C₁-C₁₂ ;

R¹ et R¹⁰ sont identiques ou différents et représentent chacun un atome d'hydrogène ou un atome d'halogène ;
R² et R⁹ sont identiques ou différents et représentent chacun un atome d'hydrogène, un atome d'halogène ou un groupe cyano, nitro, alkyle en C₁-C₁₂, halogénoalkyle en C₁-C₆, alcoxy en C₁-C₆, alkylthio en C₁-C₆, amino, mono- ou di(alkyle en C₁-C₆)amino, (alcoxy en C₁-C₆)alkyle en C₁-C₆, halogéno(alcoxy en C₁-C₆)alkyle en C₁-C₆ ou (alcoxy en C₁-C₆)carbonyle ;

R³ et R⁸ sont identiques ou différents et représentent chacun un atome d'hydrogène, un atome de fluor ou de chlore, ou un groupe alkyle en C₁-C₁₂, halogénoalkyle en C₁-C₆, halogénoalcoxy en C₁-C₆, halogénoalkylthio en C₁-C₆, halogénoalcényle en C₂-C₄, halogéno(alcoxy en C₁-C₆)alkyle en C₁-C₆, (alcoxy en C₁-C₆)carbonyle, halogénoalkylsulfinyne en C₁-C₆, halogénoalkylsulfonyne en C₁-C₆, nitro ou cyano ;

R⁴ et R⁷ sont identiques ou différents et représentent chacun un atome d'hydrogène, un atome d'halogène ou un groupe alkyle en C₁-C₁₂ ou alcoxy en C₁-C₆ ;

R⁵ représente un atome d'hydrogène, un atome d'halogène ou un groupe cyano, alkyle en C₁-C₁₂, halogénoalkyle en C₁-C₆, alcoxy en C₁-C₆, alkylthio en C₁-C₆, alkylsulfinyne en C₁-C₆ ou phényle ;

R⁶ représente un atome d'halogène ou, lorsque R⁵ est l'hydrogène, un groupe alkyle en C₁-C₁₂ ;

à condition qu'aucun des deux noyaux phényle ne soit substitué ou bien qu'au moins l'un de R³ et R⁸ soit autre chose que l'hydrogène.

2. Composé selon la revendication 1, dans lequel chacun de X¹ et X² représente un atome d'oxygène, un atome de soufre ou un groupe NH.

3. Composé selon la revendication 2, dans lequel chacun de X¹ et X² représente un atome d'oxygène.

4. Composé selon l'une quelconque des revendications 1 à 3, dans lequel

R¹ et R¹⁰ sont identiques ou différents et représentent chacun un atome d'hydrogène ou un atome de fluor ;

R² et R⁹ sont identiques ou différents et représentent chacun un atome d'hydrogène, un atome d'halogène ou un groupe nitro, alkyle en C₁-C₆ ou cyano ;

R³ et R⁸ sont identiques ou différents et représentent chacun un atome d'hydrogène, de fluor ou de chlore, ou un groupe nitro, alkyle en C₁-C₄, halogénoalkyle en C₁-C₄, halogénoalcoxy en C₁-C₄, halogénoalcényle en C₂-C₄ ou (alcoxy en C₁-C₄)carbonyle ;

R⁴ et R⁷ sont identiques ou différents et représentent chacun un atome d'hydrogène ou d'halogène ou un groupe alkyle en C₁-C₄ ;

R⁵ représente un atome d'hydrogène, un atome d'halogène ou un groupe halogénoalkyle en C₁-C₄.

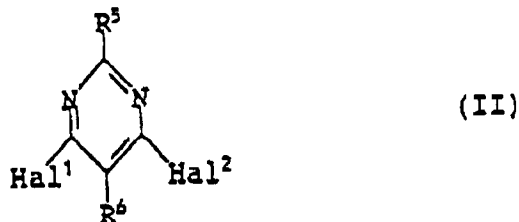
R^6 alkythio en C_1-C_4 , alkylsulfinyle en C_1-C_4 ou phényle ; et
représente un atome d'hydrogène ou, lorsque R^5 est l'hydrogène, un groupe méthyle.

5. Composé selon l'une quelconque des revendications 1 à 3, dans lequel

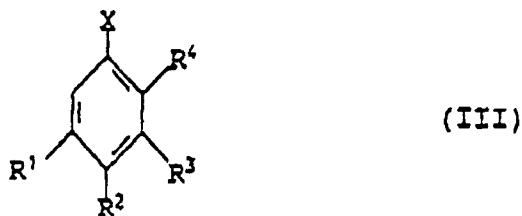
chacun de R^1 et R^{10} représente un atome d'hydrogène ;
chacun de R^2 et R^9 représente un atome d'hydrogène, de fluor, de chlore ou de brome, ou un groupe butyle, cyano ou nitro ;
chacun de R^3 et R^8 représente un atome d'hydrogène ou de chlore, ou un groupe trifluorométhyle, trifluorométhoxy, pentafluoroéthyle ou difluoroéthényle, ou bien l'un de R^3 et R^8 représente un groupe trifluorométhyle et l'autre représente un atome d'hydrogène, de chlore ou de fluor, ou un groupe méthyle, butyle, nitro, cyano ou méthoxycarbonyl ;
 R^5 représente un atome d'hydrogène, de fluor, de chlore ou de brome, ou un groupe méthyle, méthylthio, éthylthio, éthylsulfinyle ou phényle ; et
 R^6 représente un atome d'hydrogène.

6. Procédé pour la préparation d'un composé de formule générale I selon la revendication 1, qui consiste

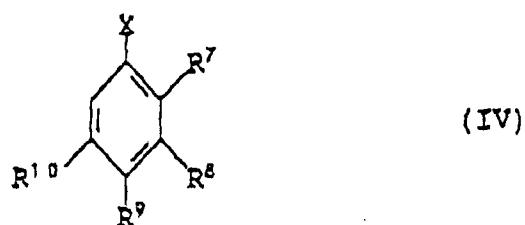
a) pour préparer des composés symétriques dans lesquels $R^1 = R^{10}$, $R^2 = R^9$, $R^3 = R^8$ et $R^4 = R^7$, à faire réagir dans des conditions basiques une 4,6-dihalogénopyrimidine de formule générale



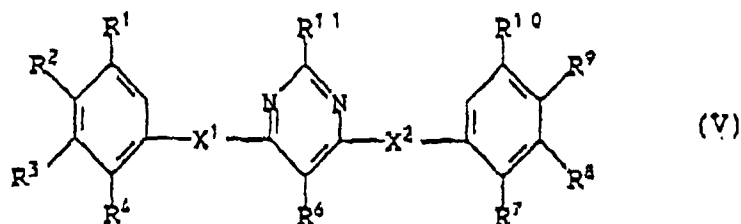
dans laquelle R^5 et R^6 sont tels que définis dans la revendication 1 et chacun de Hal^1 et Hal^2 représente indépendamment un atome d'halogène, avec un composé de formule générale



dans laquelle X représente un groupe CH_2Hal , $COHal$, OH , SH ou NRH , Hal représente un atome d'halogène, et R , R^1 , R^2 , R^3 et R^4 sont tels que définis dans la revendication 1, en un rapport molaire d'au moins 1:2 ;
b) pour préparer des composés asymétriques dans lesquels R^1 , R^2 , R^3 et R^4 ne sont pas respectivement identiques à R^{10} , R^9 , R^8 et R^7 , à faire réagir dans des conditions basiques un composé de formule II avec un composé de formule III en un rapport molaire de 1:1, puis à faire réagir le produit résultant avec un composé de formule générale



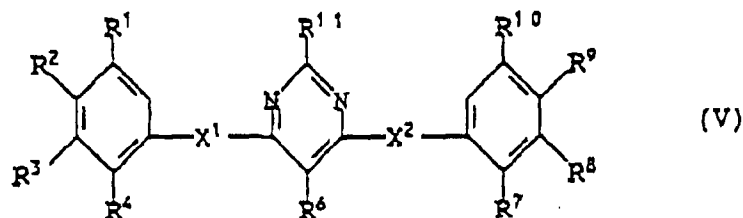
dans laquelle X, R⁷, R⁸, R⁹ et R¹⁰ sont tels que définis dans la revendication 1, également en un rapport molaire de 1:1 ; ou bien
c) à convertir un composé de formule générale



dans laquelle X¹, X², R¹, R², R³, R⁴, R⁶, R⁷, R⁸, R⁹ et R¹⁰ sont tels que définis dans la revendication 1, et R¹¹ représente un groupe OH ou NH₂, en un composé de formule générale I,

et, si cela est souhaité ou nécessaire, convertir un composé de formule générale I en un autre composé de formule générale I.

7. Composition pesticide comprenant un support et, comme ingrédient actif, un composé de formule I selon l'une quelconque des revendications 1 à 5.
8. Procédé pour combattre des nuisibles en un site, qui consiste à traiter le site avec un composé de formule I selon l'une quelconque des revendications 1 à 5, ou une composition selon la revendication 7.
9. Procédé pour combattre des ectoparasites d'animaux, qui consiste à appliquer sur la peau ou le pelage d'un animal un composé de formule I selon l'une quelconque des revendications 1 à 5, ou une composition selon la revendication 7.
10. Composé de formule générale



dans laquelle X¹, X², R¹, R², R³, R⁴, R⁶, R⁷, R⁸, R⁹, R¹⁰ et R¹¹ sont tels que définis dans la revendication 6, ou un dérivé de celui-ci dans lequel R¹¹ représente un groupe alcoxy ou un groupe mono- ou dialkylamino, à l'exception de la 2-amino-4,6-bisphénoxyypyrimidine et de la 2-amino-4,6-bis(3-chlorophénylimino)pyrimidine.